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To set up an equation in the Chemical Equations Gizmo, type the chemical formulas into the text boxes of the Gizmo. First, type in “ H_2+O_2 ” in the Reactants box and “ H_2O ” in the Products box. This...

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Equations (ANSWER ...

Chemical Equations.

Launch Gizmo. Practice balancing chemical

equations by changing

the coefficients of reactants and products.

As the equation is manipulated, the

amount of each element is shown as individual

atoms, histograms, or numerically. Molar

masses of reactants and

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products can also be calculated and balanced to demonstrate conservation of mass.

Gizmo Answers

Chemical Equations

Gizmo : Lesson Info :

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moderate aerobic activity, or 75 minutes of vigorous activity, or the equivalent combination of two during the week.

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chemical reaction, reactants. interact to form . products. This process is summarized by a chemical equation.

In the . Balancing Chemical Equations . Gizmo™, look at the floating molecules below the initial reaction: $H_2 + O_2 \rightarrow H_2O$. How many atoms are in a hydrogen . molecule (H_2)? 2 How many

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atoms are in an oxygen molecule (O₂)? 2

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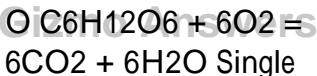
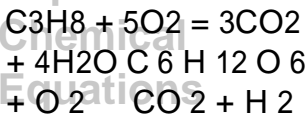
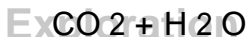
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Combustion – a fuel reacts with oxygen to release carbon dioxide, water, and heat. CH₄ + O₂ → CO₂ + H₂O

CH₄ + 2O₂ = CO₂ + 2H₂O
C₃H₈ + O₂

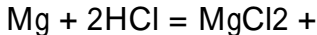
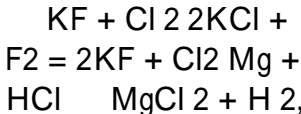
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Single replacement – an element replaces

another element in a compound. $\text{KCl} + \text{F}_2$



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H₂ Cu + AgNO₃

Cu(NO₃)₂ + Ag Cu ...

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In a chemical reaction,
reactants interact to
form products. This
process is summarized
by a chemical equation.

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Gizmo, look at the floating molecules below the initial reaction: $\text{H}_2 + \text{O}_2 \rightarrow \text{H}_2\text{O}$. 1. How many atoms are in a hydrogen molecule (H_2)? ____ 2. How many atoms are in an oxygen molecule (O_2)?

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Equation is balanced if the number of each type of atom on the left side is equal to the number of each type on the right side. Is this reaction balanced? _____

Activity A: Balancing equations Get the Gizmo ready: Check that the Synthesis reaction is selected and that all coefficients are set to one. (The

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coefficients are the numbers in the boxes.)

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1. How many atoms are in a hydrogen molecule (H_2)? ____
2. How many atoms are in an oxygen

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molecule (O₂)?

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+ O₂ H₂O. 1. How many atoms are in a hydrogen molecule (H₂)? ____ 2. How many atoms are in an oxygen molecule (O₂)?

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“H₂O” in the
Products box. This
represents the reaction
of hydrogen and oxygen
gas to form water.

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