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Answers for the questions in
Chapter 8: ROUND 3 Exercise
8-1: Using the activity
series, predict and balance
the following single
replacement reactions. Use
abbreviations to indicate
the appropriate phase of
reactants and products where
possible. Note: Not all of
the reactions will occur.
For those that do not, write
no reaction. 1.

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the questions in Chapter 8:
ROUND 3 Exercise 8-1: Using
the activity series, predict
and balance the following
single replacement
reactions. Use abbreviations
to indicate the appropriate
phase of reactants and
products where possible.

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...

The Ultimate Chemical
Equations Handbook Determine
the oxidation number of the
underlined element: KMnO_4

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Since K is an alkali metal, its charge must be $1+$ Oxygen is $2-$ but there are four of them, therefore, 4 times $2-$ equals $8-$ If $1+$ and $8-$ are added Ultimate Chemical Equation Answers

The Ultimate Chemical
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Chapter 7

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many of the transuranium
elements were prepared at
the Lawrence Radiation
Center, University of
California at Berkeley,
California, U.S.A., it is
logical to find the
following names:
Americium—named after the

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Americans (after all, one must be patriotic!).

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Quiz on all Chemical

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Pg. 51 • Acid Base
Neutralization Reactions-
Pg. 53, Practice-Pg. 54 • Weak
acids and bases do not break
apart and are in net ionic
equation. • Synthesis and
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School of Texas, Dallas, TX,
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